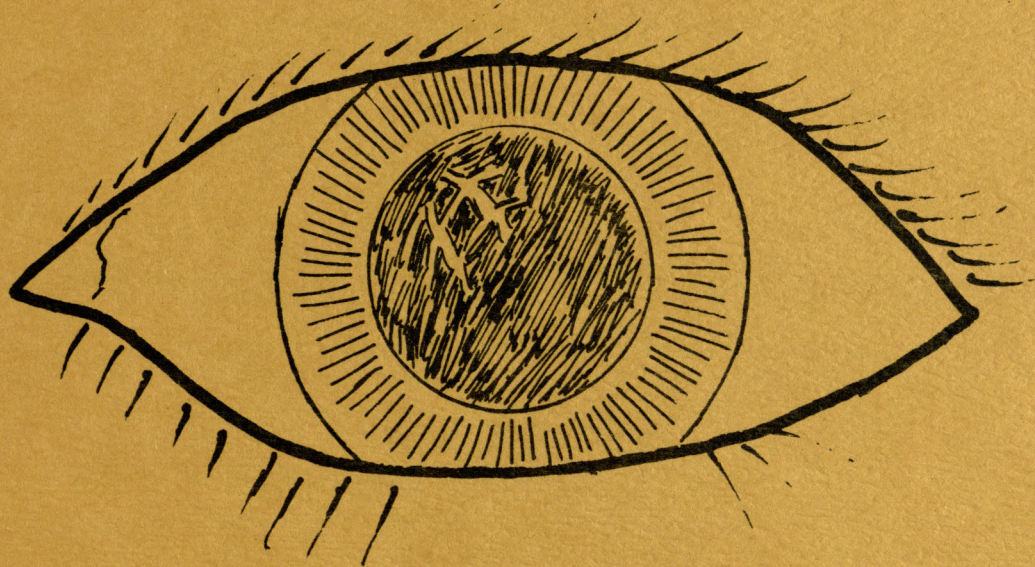


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MY EYE

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Ka-a! Iiska!



Aqnightuq!



(2)



Mingugpenaan

(3)



Utaqigi. Iyegpek iqitesnaagha.

(4)

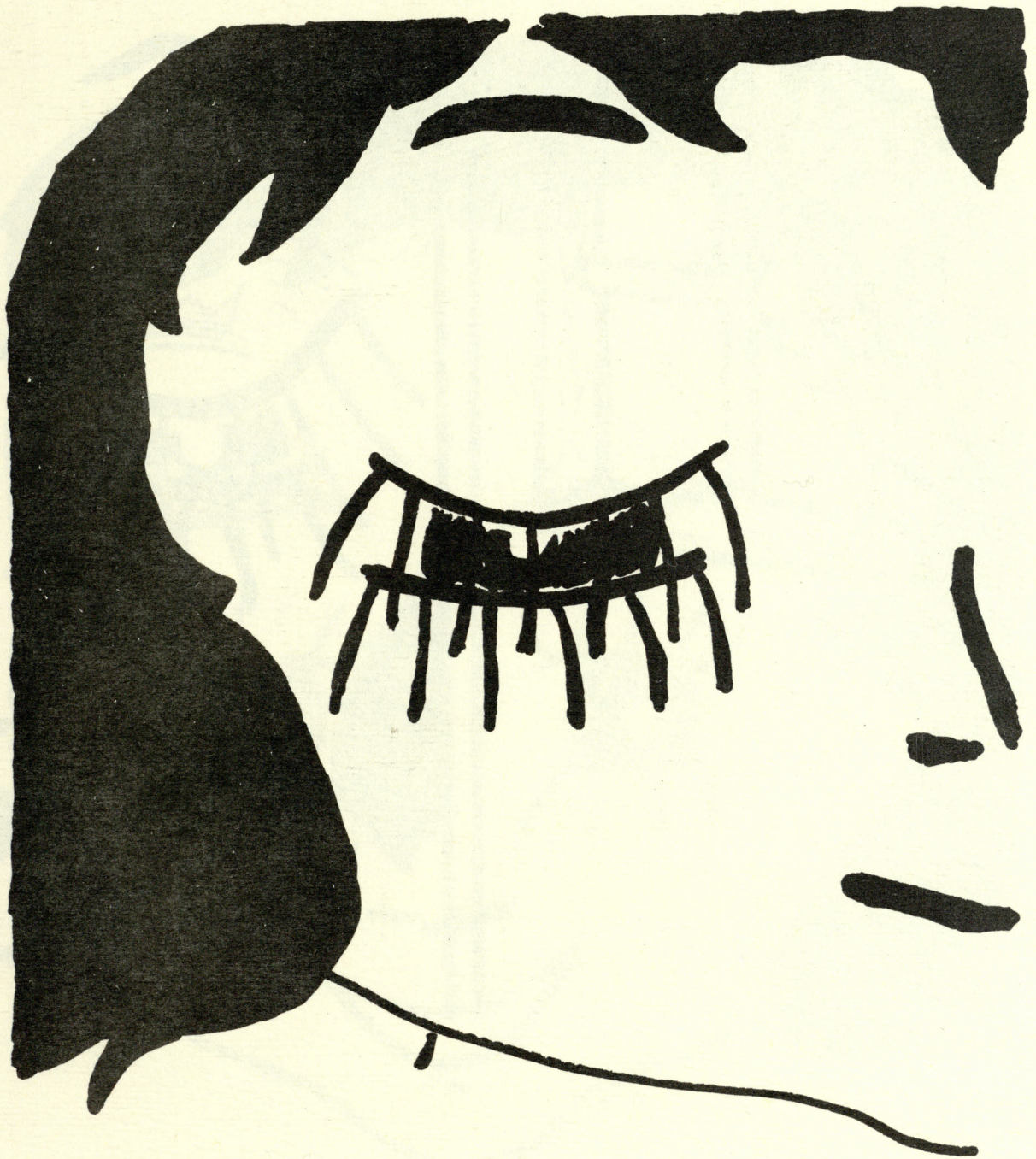
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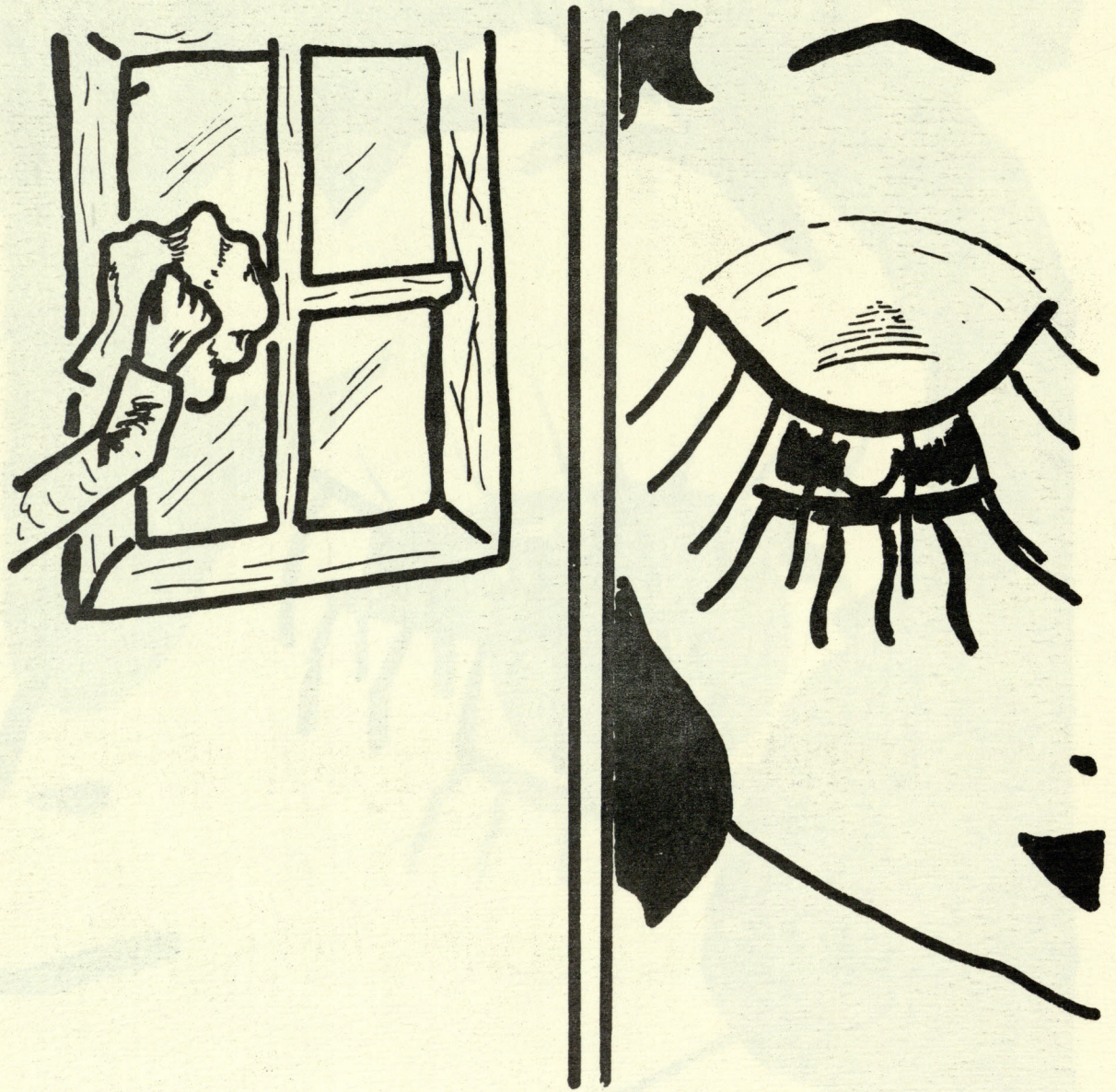


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mallat kiyaghtestaqi iyegpeneng.
(7)

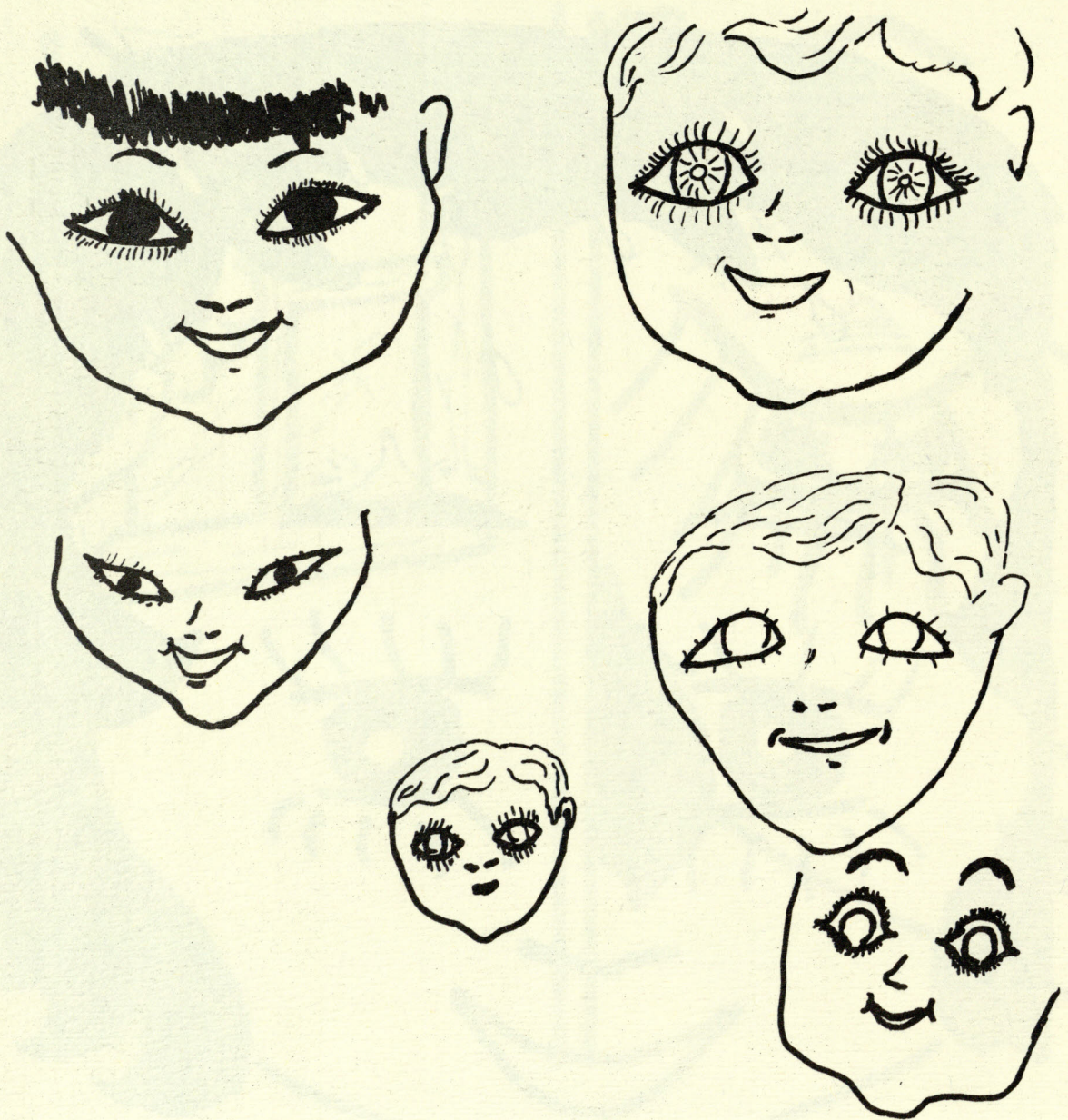


Elpek iyegpek qulegkenka
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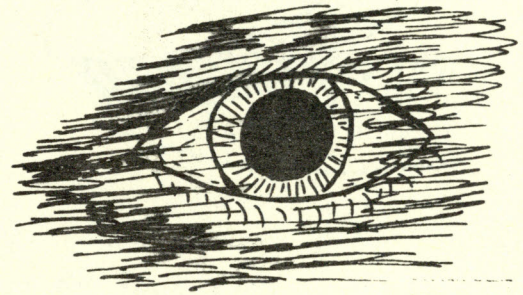
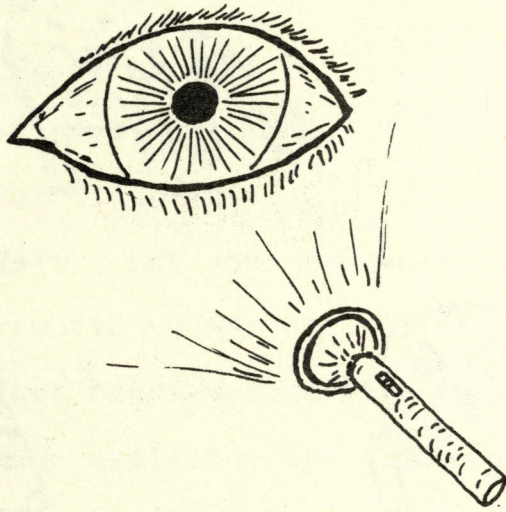
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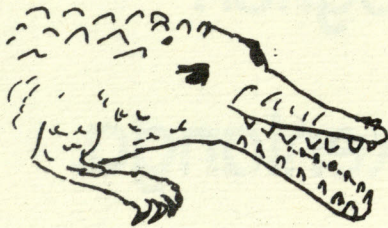
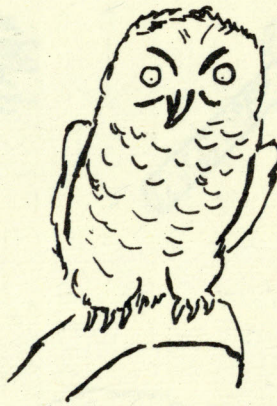
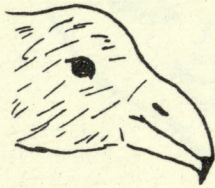
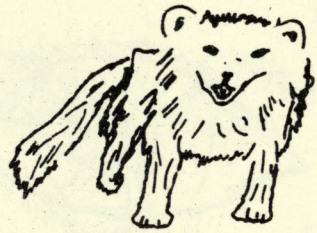
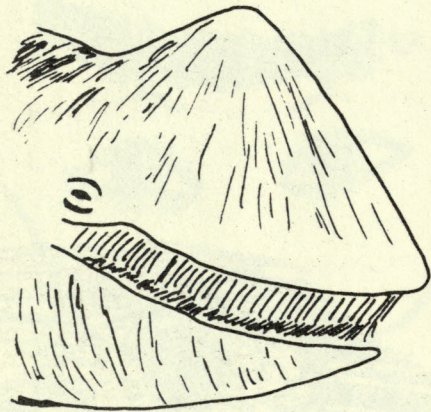
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Quneghllukek iiyegken
kesti quilmun kenlanga.

Qineghten

taghneghaghqutegnun.



Iyet

(12)

EYES

1. O-o-oh my eye!
2. It hurts!
3. Don't rub it!
4. Wait. Let your eye work it out.
5. How did my eye get rid of it?
6. Your tears washed it out.
7. Your eyelashes help keep dust out of your eyes.
8. Your eyelids wipe the eye clean.
9. Your eyebrows keep water from rolling down your forehead.
10. Eyes
11. Close your eyes and count to ten. Look in a mirror.
12. Eyes



1. The first part of the report is a general
 introduction to the subject of the study.
 2. The second part is a detailed description
 of the methods used in the investigation.
 3. The third part contains the results of the
 experiments and a discussion of their
 significance.
 4. The fourth part is a summary of the
 findings and conclusions of the study.
 5. The fifth part is a list of references
 cited in the report.

The following table shows the results of the
 experiments conducted under various conditions.
 It is evident that the rate of reaction
 increases with increasing temperature and
 decreasing concentration of the reactants.
 This is in accordance with the law of mass
 action and the Arrhenius equation.

In conclusion, the study has shown that the
 rate of reaction is affected by temperature
 and concentration. The results are in
 agreement with the theoretical predictions.
 Further studies are needed to determine the
 effect of other factors on the reaction rate.



